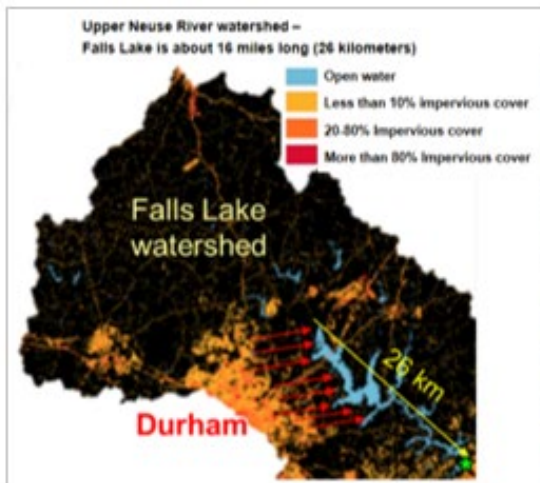


Summer Program Materials

Handout – Introduction to Falls Lake and its Watershed (2 pp.)

(also explains how to assess lake health using background environmental conditions, water quality, and biota characteristics.)

FALLS LAKE – most important drinking (potable) source-water for the City of Raleigh, which supplies water to Garner (actually an artificial or human-constructed lake, also called a reservoir or an impoundment)

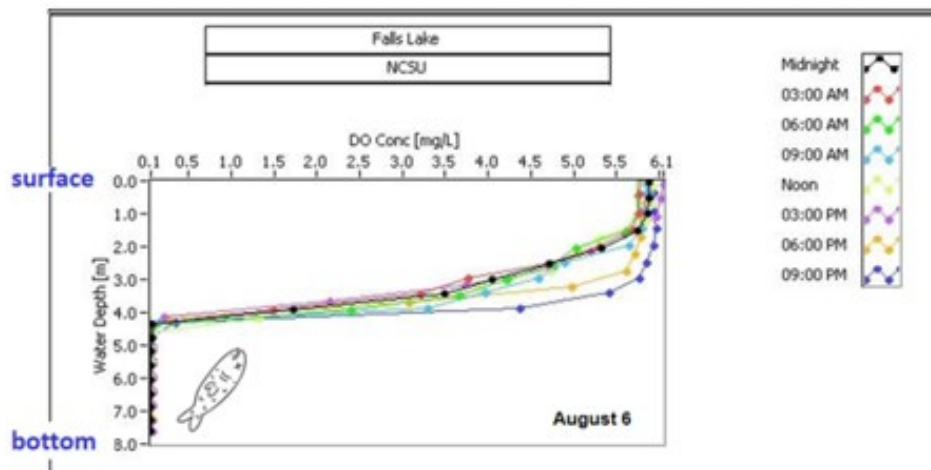


Assessing aquatic ecosystem health

Background environmental conditions

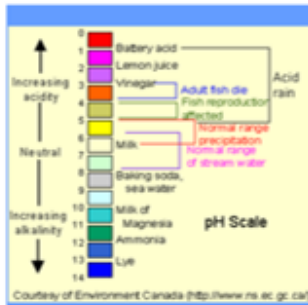
Water temperature – hotter or colder than normal stresses beneficial aquatic life

Oxygen – produced by algae and plants, and diffuses in from the air; worst conditions at depth, and just before dawn (“sag”)



$\text{pH} = -\log [\text{H}^+] =$ minus the \log_{10} of the hydrogen ion (H^+) concentration;
exponential (logarithmic or log-base-10 scale):

Log scale - a small change in pH is equal to a **LARGE** change in acidity



Acidic = less than 7

Neutral = 7

Basic = greater than 7

pH – **master variable** for the chemical state of an aquatic ecosystem because it is a reliable indicator of whether the system is able to protect or **buffer** itself from sudden acidic or basic pollution; and because almost any process (physical, chemical or biological) affects pH either directly or indirectly.

Water quality indicators – example, cultural eutrophication:

Nutrient (N, P) concentrations and algae as “first responders”
(biomass or abundance, indicated by the chlorophyll a concentration)



high nutrient pollution
↓
high algal biomass
(indicator, chlorophyll a)

Other biological indicators (higher trophic levels in the food web)

